(SBL) SH
$$\frac{pH 9.5}{CH_3 - SO_2 - S - R}$$
 (SBL) S R

$$R = \frac{1}{CH_3 - SO_2 - S - R}$$

$$A, X = H$$

$$b, X = CH_3$$

$$c, X = OCH_3$$

$$d, X = COOH$$

$$e, X = F_5$$

$$f, X = CF_3$$

$$g, X = 2, 4 - diNO_2$$

F1G. 1.

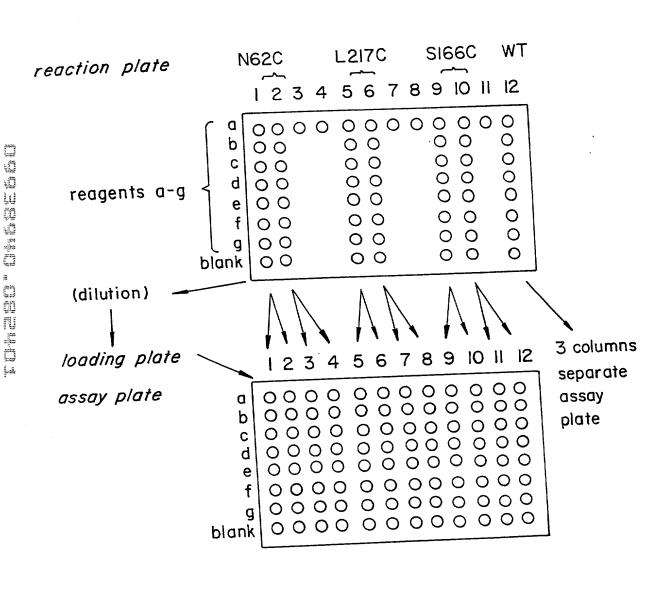
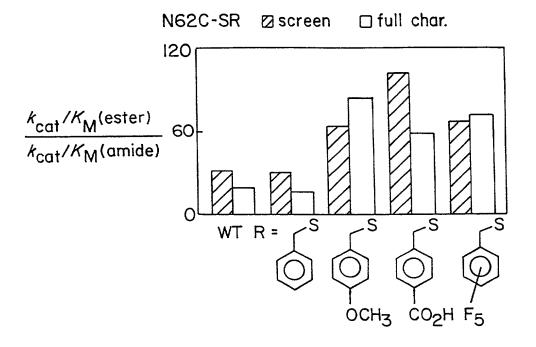


FIG. 2.

FIG. 3.



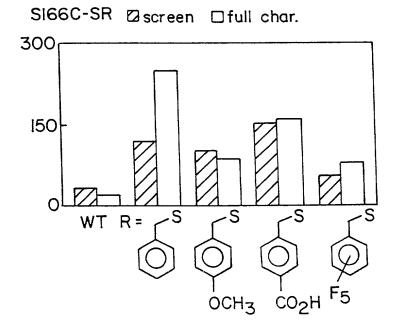


FIG. 4.

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